

## **SYLLABUS: CHEMISTRY OF CONTINENTAL WATERS W4885y Spring '04**

The purpose of this course is to outline some aspects of the chemistry of natural waters on the continents, including rain, snow, springs, small streams, rivers, lakes, groundwaters and estuaries. It is directed to undergraduate & graduate students. The chemistry of many natural waters have been affected by human activities. Our focus will be on both anthropogenic effects and natural processes. Attention will be given to properties of water, aqueous chemical equilibria, acidity, alkalinity & pH, oxidation-reduction reactions and chemical weathering. Discussion will approximately follow the natural hydrologic cycle from formation of rain and snow to weathering of rocks in soils, followed by transport in rivers, temporary retention in fresh water lakes, or groundwaters and finally saline lakes & estuaries. Emphasis will be placed on integrating information from water budgets, chemical equilibria concepts, major and trace element compositions, and distributions of natural and anthropogenic radionuclides. Case studies from a number of geographic regions, with practical management implications, will be discussed to illustrate processes which affect the chemistry of a range of types of continental waters.

Primary Text: Global Environment, Water, Air, and Geochemical Cycles, E. K. Berner and R. A. Berner, Prentice-Hall, Upper Saddle River, NJ 07458, 1996, 376 pp (paperback).

Suggested additional resource: Biogeochemistry of a Forested Ecosystem, 2nd Edition, G. E. Likens and F. H. Bormann, Springer-Verlag, New York, 1995, 159 pp (paperback). Both should be available from Labyrinth Bookstore (112th St). About 75% of Berner & Berner plus all of Likens & Bormann will be assigned reading during the course.

Problem Sets, Midterm & Final Exams

Tuesday/Thursday 10:00 -11:15 AM, 558 Schermerhorn Annex

Telephone calls from main campus to LDEO: 95-xxxx

Instructors: Robert Anderson, LDEO (845) 365-8508 or (845) 353-2165 (home)

James Simpson, LDEO (845) 365-8570 or (845) 358-1613 (home)

Grad Teaching Assistant, Alison Keimowitz, LDEO (845) 365-8510, Seismology Rm #214A

e mail addresses: boba, simpsonj, ark @ldeo.columbia.edu

Office hours arranged by appointment

## GENERAL OUTLINE OF TOPICS

The items below provide a general range of relevant topics, rather than a list of specific lecture titles. Most of these topics will be treated to some degree during the semester.

- I. Atmospheric Water Cycle and Precipitation Chemistry
  - Chemical bonding of water molecules and related physical properties
  - Evolution of planetary atmospheres & structure of Earth's atmosphere
  - Atmospheric chemistry: aerosols
  - Radioactive fallout as tracers
  - Marine aerosol and precipitation chemistry (Hawaii)
  - Continental precipitation chemistry, including S & N cycles (NE USA)
  
- II. Properties of Water, Weathering Reactions, Spring & Stream Chemistry
  - Chemical equilibria in aqueous solutions & acid-base aqueous chemistry
  - Solid phase & solution exchange reactions
  - Redox equilibria
  - Weathering of granitic and volcanic rocks (Sierra Nevada & Mt. Mazama, OR)
  - Major element and radionuclide budgets in lakes (Crater Lake, OR)
  - Forested ecosystems (Black Rock Forest, NY & Hubbard Brook, NH)
  - Chemistry of rivers: Amazon, Nile, Murray (Australia)
  
- III. Biogeochemical Budgets in Freshwater Lakes
  - Physical and biogeochemical processes
  - Gas exchange between surface waters and the atmosphere
  - Photosynthesis, respiration and redox chemistry
  - Sediment pore water chemistry
  - Eutrophication (Experimental Lakes Area, Ontario)
  - Early effects of global warming on lakes
  - Trace metal cycles
  - Particle dynamics and sedimentation
  - Adsorption and "scavenging" - natural cleansing processes
  
- IV. Groundwaters, Saline Lakes and Estuaries
  - Rockland County, NY & Murray Basin, Australia
  - Arsenic mobilization in reducing groundwaters
  - Closed basin lakes (Mono Lake, CA & the Dead Sea in Israel)
  - Meromictic lakes
  - Estuary physical transport processes
  - Nutrient & dissolved oxygen cycles as impacted by sewage influxes (Hudson)
  - Non conservative behavior of trace elements in estuarine dissolved transport
  - Persistent contaminants: radionuclides, trace metals & PCBs (Hudson River)

**ASSIGNMENT W4885 January 20, 2004**

Text (Berner & Berner)

Chapters 1, 2, 3

1. Introduction: the water cycle, atmospheric and oceanic circulation
2. Air chemistry: the greenhouse effect and the ozone hole
3. Rainwater and atmospheric chemistry

Read the following pages as directly relevant preparation for this course:

1 - 9, 48 - 56, 62 - 130

The following pages should be skimmed if you are not already familiar with the material:

9 - 24, 27 - 48

## Chemistry of Continental Waters (W4885) Spring '04

List of Lecture Titles: revised (2/21/04)

Topics covered on first exam: Lectures #1-15

Lect #	(Date)	Title
1	(1/20)	Properties of water; Hydrogen Bond; abundance of H <sub>2</sub> O
2	(1/22)	O <sub>2</sub> evolution; atmos H <sub>2</sub> O cycle & T structure; atmos circulation
3	(1/27)	Formation of aerosols; sea salt nuclei & precipitation chemistry
4	(1/29)	Hawaii rain chemistry; cont. precip chem; S gases; acid emissions
5	(2/3)	S & N oxidation states; N gases; acid emissions & deposition
6	(2/5)	West Point, NY precipitation chemistry: acid rain
7	(2/10)	Weathering reactions: evaporites, sulfide oxidation, Sierra Nevada granites
8	(2/12)	Soils, weathering of silicates; organic acids; Sierra Nevada granites
9	(2/17)	Large river TDS: climate & topography; Amazon chemical weathering
10	(2/19)	Introduction to NYC water supply system
11	(2/24)	NYC water supply system operations and monitoring
12	(2/26)	Large river TSS, Tarbela Dam sedimentation & seepage water chem
13	(3/2)	Hudson R. dissolved O <sub>2</sub> , nutrients, persistent chlorinated hydrocarbons
14	(3/4)	River and groundwater chemistry in Australia
15	(3/9)	Arsenic and other groundwater contaminants from landfills
16	3/11)	<b>MIDTERM EXAM</b>
		<b>SPRING BREAK</b>
17	(3/23)	General chemical equilibrium relationships
18	(3/25)	Complexation & solubility equilibria
19	(3/30)	Exchange equilibria with solid surfaces #1
20	(4/1)	Exchange equilibria with solid surfaces #2
21	(4/6)	Acid-base equilibria #1
22	(4/8)	Acid-base equilibria #2
23	(4/13)	Oxidation-reduction equilibria
24	(4/15)	Classification of micro-organisms; biological processes, redox reactions
25	(4/20)	Hubbard Brook Experimental Forest #1
26	(4/22)	Hubbard Brook Experimental Forest #2
27	(4/27)	Lakes #1
28	(4/29)	Lakes #2
	(5/xx)	<b>EXAM</b>