## MEASURING LOW-TEMPERATURE CO2-H2O EQUILIBRIA: START-UP

P.I. John Longhi, D.S.R.S.

June 30, 2003

## **ABSTRACT**

Vast potential sinks for anthropogenic CO2 exist within and possibly beneath the Antarctic and Greenland ice sheets. In the process of working out the phase equilibria in the low-temperature portion of the CO<sub>2</sub>-H<sub>2</sub>O binary system relevant to Mars' volatile history (the atmospheric composition, which is 95 % CO<sub>2</sub> and 0.03% H<sub>2</sub>O, is buffered by water ice), it became apparent to me that the CO<sub>2</sub>-clathrate phase (CO<sub>2</sub>•5.75H<sub>2</sub>O) is thermodynamically stable in the presence of water ice over a most of the pressure-temperature range of the terrestrial ice sheets. Furthermore, there are significant portions of the ice sheet P-T range over which liquid CO<sub>2</sub> is stable either with clathrate or liquid water. There is relatively little data on the mutual solubility of CO<sub>2</sub> and H<sub>2</sub>O at the relevant equilibrium conditions, so I am requesting funds to begin the process of setting up an experimental apparatus that will enable measurement of the various equilibria and phase composition.